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Investigation of the optical and excitonic properties of the visible-light driven photocatalytic BiVO₄ material

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ABSTRACT: An investigation of the optical and excitonic properties of photocatalytic compounds is proposed based on both experimental and theoretical approaches. More specifically, this paper reports, for the very first time, on the local-field, optical anisotropy and excitonic effects in BiVO₄, an active photocatalytic material in the visible range. The analyses are based on electron energy loss spectroscopy measurements, ground-state DFT calculations including crystal local-field effects and many-body corrections using the Bethe Salpeter equation. These results are supported by a comparison with two materials, namely TiO₂ anatase and rutile, which are well known to differ in their photocatalytic properties, being important and negligible, respectively. The analogies found for these two categories of compounds allow proposing criteria that appear essential to obtain an optimal photocatalytic material.

INTRODUCTION

In 1972, Honda and Fujishima first discovered that it was possible to promote the water splitting reaction, i.e. the water decomposition into oxygen and hydrogen, by using an electrochemical device based on a TiO₂ anode and a platinum counter electrode exposed to UV-light.¹ Over the past four decades, an important number of investigations has been devoted to such photoelectrochemical (PEC) reactions. Namely, TiO₂ is well-suited for photocatalytic applications due to its low cost, its high stability in aqueous solutions and the right positioning of its valence band maximum (VBM) and conduction band minimum (CBM). Unfortunately, due to its too large band gap (> 3 eV), it operates only in the UV range and thus considerably reduces the efficiency of the solar to hydrogen conversion. To have efficiencies above 15% a band gap below 2.2 eV is required². One strategy to reach this band gap requirement is to consider multiterinary oxides combining nd^o (e.g. Ti⁴⁺, V⁵⁺, Nb⁵⁺) and ns² cations (e.g. Bi³⁺, Sn²⁺). Then, energy levels can be interspaced between the former valence and conduction bands and thus favor the absorption in the visible range. In that context, bismuth vanadate (BiVO₄), initially known for its ferroelastic and ionic conductivity properties^{3,4,5}, is now considered as one of the most promising candidate for solar-to-hydrogen conversion processes operating in the visible light

range^{6,7,8}. However, its properties strongly depend on the allotropic form of BiVO₄. Three main polymorphs are known, namely the zircon structure with tetragonal symmetry (Z_T-BiVO₄) and two (almost identical) scheelite structures with monoclinic (S_M-BiVO₄) and tetragonal (S_T-BiVO₄) symmetries, the former being ferroelastic, the latter paraelastic⁹. S_M- and S_T-BiVO₄ were selectively prepared by many groups showing that S_M-BiVO₄ exhibits high photoelectrochemical (PEC) properties, whereas the photoactivity of Z_T-BiVO₄ is negligible^{6,7,9}. It has also been demonstrated that both S_M- and S_T-BiVO₄ structures are photocatalytic active for the degradation of rhodamine B in solution under visible light, but not Z_T-BiVO₄.¹⁰

Recently, H. Fan *et al.*¹¹ shed light on the higher PEC activity of S_M-BiVO₄ compared to Z_T-BiVO₄. Based on surface photovoltage and transient photovoltage measurements under visible light irradiation, the authors clearly evidenced that photo-induced electrons (holes) flow from the surface (bulk) to the bulk (surface) for Z_T-BiVO₄ while the opposite is observed for S_M-BiVO₄. This explains why S_M-BiVO₄ is more active than Z_T-BiVO₄ for oxidation reactions. In addition, R. Li *et al.*¹² have evidenced that, under photo-irradiation, efficient charge separation is achieved in S_M-BiVO₄, leading to a reduction reaction with photo-generated electrons and oxidation reaction with photo-

generated holes on the $\{010\}$ and $\{110\}$ crystal facets, respectively.

In order to address the differences between S_{M^-} , S_{T^-} and Z_{T^-} - BiVO_4 the optical properties have been investigated. Up to date, only the optical band gaps of S_{M^-} , S_{T^-} and Z_{T^-} - BiVO_4 have been estimated based on diffuse reflectance spectra on powder samples, and no experimental investigation on their frequency dependent dielectric function (DF) has been reported so far. From the theoretical side, a significant number of studies has been carried out using density functional theory (DFT) with the generalized gradient approximation (GGA) and hybrid functionals. However, contradictory results have been published. For instance, the band gap nature of S_{M^-} - BiVO_4 was reported both as direct^{2,13} and indirect^{14,15}. Moreover, two different groups have reported simulations of the frequency dependent dielectric function^{14,15}, suggesting that the optical anisotropy is strong in S_{M^-} - BiVO_4 , and negligible in Z_{T^-} - BiVO_4 ¹⁴.

Clearly, all these previous studies of the optical properties of BiVO_4 suffer from being purely experimental or theoretical. An investigation of the optical and excitonic properties of BiVO_4 combining experiments and theory in order to raise our understanding of the photocatalytic activity of this material is still missing. This is, indeed, the main purpose of the present work, which reports, for the very first time, on the local-field, optical anisotropy and excitonic effects in S_{M^-} and Z_{T^-} - BiVO_4 , based on electron energy loss spectroscopy (EELS) measurements, ground-state DFT calculations including crystal local-field effects¹⁶ (LFE) and many-body corrections using the Bethe Salpeter equation¹⁷ (BSE). An excellent agreement is obtained between EELS experiments and DFT calculations, which fully validates such *ab initio* calculations as a starting point for more sophisticated calculations devoted to the excitonic properties. In particular, solving the Bethe-Salpeter equations (BSE) allows studying the spatial distribution of the photo-induced electron-hole pairs, i.e. the excitons, in S_{M^-} and Z_{T^-} - BiVO_4 phases, and evidences the impact of the long-range crystalline structure on it. From the present work, a unified picture of the optical and excitonic properties of S_{M^-} and Z_{T^-} - BiVO_4 is proposed which emphasizes the link between the nature of band gap, the photocatalytic activity, and the excitonic properties. Finally, this result is supported by a comparison of two materials, namely TiO_2 anatase and rutile, which are well known for their differences in the photocatalytic property, being important and negligible, respectively.

RESULTS AND DISCUSSION

Structurally speaking, S_{M^-} and S_{T^-} - BiVO_4 are very similar¹⁸, except a small distortion, which reduces the symmetry from tetragonal (space group I_{4_1}/a for S_{T^-} , with $a = b = 5.147$ and $c = 11.7216$ Å) to monoclinic (space group I_{2_1}/b for S_{M^-} , with $a = 5.1935$, $b = 5.0898$ and $c = 11.6972$ Å). In contrast, the zircon phase¹⁹ (Z_{T^-} - BiVO_4 - space group I_{4_1}/amd , with $a = b = 7.303$ and $c = 6.584$ Å) is significantly different from the two scheelite phases, leading to a large volume expansion to 87.8 Å³/f.u. compared to the S_{M^-} and

S_{T^-} phases, with a volume of 77.6 and 77.3 Å³/f.u., respectively. Using the following setting, i.e. I_{4_1}/amd , I_{4_1}/a and I_{2_1}/b space groups, for Z_{T^-} , S_{T^-} and S_{M^-} - BiVO_4 , respectively, it is possible to compare these three allotropes, with a common definition of the c -axis and the (a,b) -plane. Such a setting will be used all along the manuscript, allowing to discuss optical anisotropy and excitonic spatial distribution using a common setting for the crystallographic directions for the BiVO_4 phases. Both scheelite (S_{T^-} and S_{M^-}) and zircon (Z_{T^-}) structures consist of isolated $[\text{VO}_4]^{3-}$ tetrahedra connected by $[\text{BiO}_8]^{5-}$ antiprisms but differ in the way these polyhedra are connected to each other. The structural phase transition between zircon and scheelite has been evidenced to be based on a first-order reconstruction involving a “bond-switching” mechanism.²⁰ Figures 1a and 1b show the consequences of such a structural filiation between the S_{M^-} (or S_{T^-}) and Z_{T^-} - BiVO_4 phases. Starting from the S_{M^-} phase, three changes are observed to reach the Z_{T^-} phase: (1) a rotation of the $[\text{VO}_4]^{3-}$ tetrahedral units; (2) an expansion along the a - and b -axes of the monoclinic cell; (3) a strong compression along the c -axis. The main impact of these structural modifications concerns the interconnection between the $[\text{VO}_4]^{3-}$ and $[\text{BiO}_8]^{5-}$ units as shown in Figures 1c and 1d. The Bi-V bond lengths and Bi-O-V bond angles of S_{M^-} and Z_{T^-} - BiVO_4 are summarized in Table S1 and will be discussed in more details below, in the section devoted to the optical anisotropy.

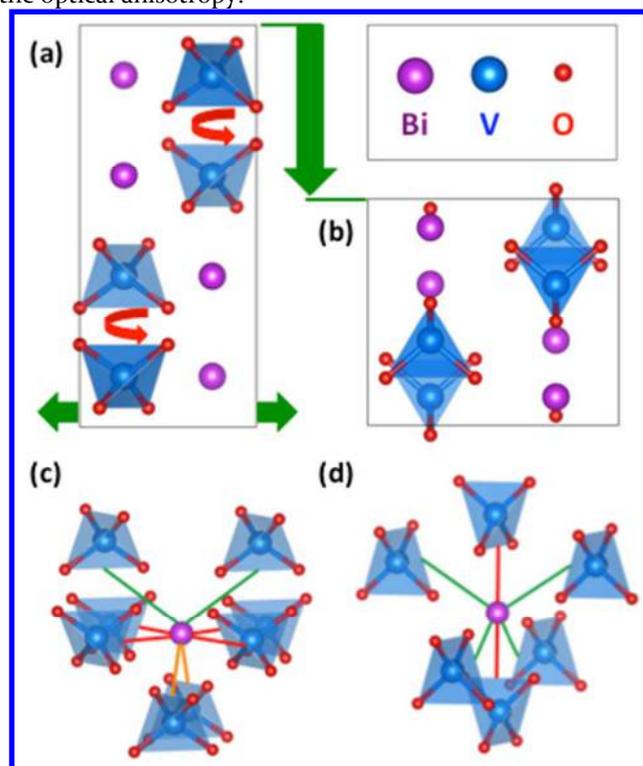


Figure 1. Schematic representation of the atomic structures of S_{M^-} - BiVO_4 (a) and Z_{T^-} - BiVO_4 (b). The structure filiation from S_{M^-} and Z_{T^-} - BiVO_4 is highlighted with the green arrows (cell expansion along a and b , cell contraction along c) and the red arrows (rotation of VO_4^{3-} tetrahedra). The resulting local environments of bismuth atoms are given for S_{M^-} - BiVO_4 (c)

and $Z_T\text{-BiVO}_4$ (d). A color code for the Bi-V bonds is used: shortest in red, intermediate in orange, largest in green (see Table S1 for the related values).

Before discussing our results on the optical anisotropy of these compounds, the properties of their excitons and their potential impact on the related photocatalytic properties, we first validated our theoretical approach. To do so, we have compared the calculated energy loss-functions with the measured low-loss EELS spectra obtained from S_M and $Z_T\text{-BiVO}_4$ samples. The Density Functional Theory (DFT) calculations have been carried out by using two different codes: the Vienna Ab initio Simulation Package (VASP)²¹ for the energy-loss functions including LFE and the WIEN2k program package²² as starting point for BSE evaluations of the excitonic properties. Most DFT calculations were performed with the Perdew, Burke, and Ernzerhof parameterization of the GGA (PBE-GGA) of the exchange-correlation potential.²³ BSE calculations were performed using a similar strategy than the one we used for CuAlO_2 compound.²⁴

ENERGY-LOSS FUNCTION OF S_M - AND $Z_T\text{-BiVO}_4$

Electron energy loss spectroscopy (EELS) measurements are directly related to the energy-loss function (ELF) defined by $\text{Im}[-1/\epsilon(\nu)] = \epsilon_2(\nu) / (\epsilon_1(\nu)^2 + \epsilon_2(\nu)^2)$, where $\epsilon_1(\nu)$ and $\epsilon_2(\nu)$ are the real and imaginary parts of the dielectric function, where ν is the frequency²⁵. Properly combined with first-principles calculations, it is an unrivaled tool to probe the frequency dependent dielectric function over a high-energy range.²⁶ The optical diffusion and absorption processes are directly related to $\epsilon_1(\nu)$ and $\epsilon_2(\nu)$, respectively. Figure 2 shows the energy loss function, $\text{Im}(-1/\epsilon)$, of S_M and $Z_T\text{-BiVO}_4$. The experimental measurements have been carried out with a low collection angle, allowing a direct comparison between the recorded and calculated loss-functions at zero momentum transfer ($q \rightarrow 0$). An excellent agreement is obtained between the present DFT calculations and orientation-dependent EELS measurements, as soon as the local-field effects (LFE) are taken into account in our simulations (more details on LFE are provided in the supplementary materials). In particular, the energy position and respective intensity each spectral feature is well-reproduced. Such an agreement between PBE-GGA calculations and EELS measurements is not expected at a first sight. Indeed, we expect an underestimation of the band gap value using PBE-GGA and thus a shift of all the simulated peaks with respect to the experimental spectrum. However, as previously reported,²⁶ PBE-GGA simulations give correct peak positions for the high-energy part of the spectrum ($E > 5$ eV), while the band gap underestimation has a significant impact on the low-energy part of the spectrum ($E < 5$ eV). We are in a similar situation except that no significant spectral feature is detectable at low-energy, explaining such an overall agreement. In addition, it should be noticed that PBE-GGA underestimates the band gap of BiVO_4 compounds by only 0.4 eV (see below).

We should mention that the loss-functions are nearly the same along the three crystallographic directions for

$S_M\text{-BiVO}_4$, in contrast to $Z_T\text{-BiVO}_4$. This observation is important and allows us to consider an average loss-function for $S_M\text{-BiVO}_4$, which makes the results independent of the orientation chosen during the experiments. In contrast, the comparison for $Z_T\text{-BiVO}_4$ requires a careful definition of the orientation axis (the origin of the different features is explained in Fig. S2 in the Supplementary Materials). It also should be noted that the tail of the zero-loss peak hinders the proper extraction of information below ~ 5 eV (the energy resolution of the microscope is around 0.8 eV, see Supplementary Materials). As a consequence band-gap offset and excitons cannot be probed with this experimental set-up.

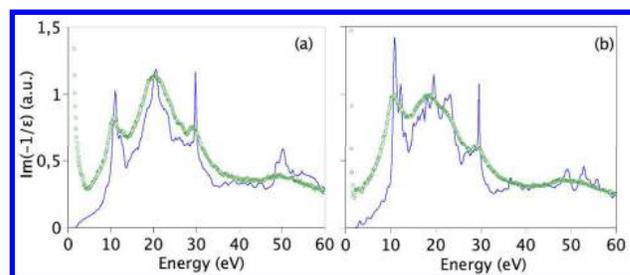


Figure 2. Energy-loss ($\text{Im}(-1/\epsilon)$) functions of $S_M\text{-BiVO}_4$ (a) and $Z_T\text{-BiVO}_4$ (b). The experimental VEELS spectra (green open circles) of S_M - and $Z_T\text{-BiVO}_4$ have been recorded along the $[9\ 11\ 4]$ and $[1\ 1\ 0]$ zone axes, respectively. The calculated energy-loss and dielectric functions include local-field effects. For comparison, the calculated quantities correspond to an average of the polarizations along x , y and z for $S_M\text{-BiVO}_4$ and a polarization along z for $Z_T\text{-BiVO}_4$.

OPTICAL ANISOTROPY IN BiVO_4 AND TiO_2

Using a standard DFT functional appears adequate for the description of the high-energy range of the dielectric function (from 5 to 60 eV). However, since band gaps are strongly underestimated in conventional DFT, a more advanced method is needed for an accurate description of the low-energy range (from 0 to 5 eV for instance). Consequently, we have used the Tran-Blaha modified Becke-Johnson (TB-mBJ) potential,²⁷ which is known for its ability to describe band gaps with an accuracy comparable to sophisticated GW calculations. In our DFT calculations we have also taken into account relativistic effects on the valence electrons by including the spin-orbit (SO) coupling in our DFT calculations, to properly describe the $\text{Bi-p}^{1/2}$ and $\text{Bi-p}^{3/2}$ states, which are involved in the bands close to the Fermi level. Indeed, a band gap reduction of about 0.1 eV is observed when the SO coupling is switched on, for both S_M - and $Z_T\text{-BiVO}_4$.

Since in optical excitations the excited electron may interact with the hole it left behind, we had to go beyond standard DFT. By solving the 2-particle Bethe Salpeter Equation (BSE) one can properly describe the interaction between the electron and the hole of a photo-generated pair, i.e. the bound exciton state. Here, the starting point of the BSE calculations was not based on GW eigenvalues but on our GGA results plus a band gap correction, which was defined in such a way to reproduce the TB-mBJ and

GW band gaps of BiVO₄ and TiO₂, respectively. Such an approximation of the BSE spectra leads to similar results for TiO₂ than the ones starting from GW eigenvalues³¹.

From our DFT calculations (including SO coupling) we obtain an indirect and direct fundamental band gap for S_M- and Z_T-BiVO₄, which changes from 2.06 and 2.47 eV to 2.46 and 2.84 eV for the two phases respectively, when the TB-potential mBJ is taken into account. This leads to a band gap correction of about 0.4 eV for the BiVO₄ compounds.

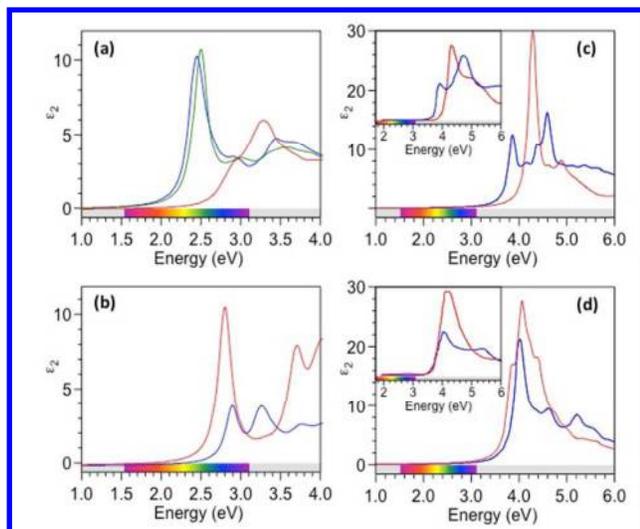


Figure 3. Imaginary part of the dielectric function for S_M-BiVO₄ (a), Z_T-BiVO₄ (b), ana-TiO₂ (c) and rut-TiO₂ (d) calculated via Bethe-Salpeter equation (BSE). The BSE calculated spectra have been carefully converged for the energy range shown here (i.e. up to 4 and 6 eV for BiVO₄ and TiO₂, respectively). The x, y and z polarizations are represented with blue, green and red solid lines. Insets: Experimental spectra for anatase-TiO₂ and rutile-TiO₂.

It should be noticed that the indirect nature of the band gap of S_M-BiVO₄ has been recently experimentally confirmed,²⁸ with an energy difference between the indirect (2.56 eV) and direct (2.68 eV) band gaps of 0.12 eV, which agrees quite well with our prediction of 0.12 eV (2.47 and 2.59 eV, respectively for indirect and direct band gaps for S_M-BiVO₄). In contrast, our calculations show a difference of 0.02 eV between the indirect (2.85 eV) and the direct (2.87 eV) band gaps for Z_T-BiVO₄, which is not significant.

In Figs. 3a and 3b, the ε₂ function (in the low-energy range), deduced from BSE calculations, is plotted for S_M- and Z_T-BiVO₄, respectively. For comparison we also show (in Figs. 3c and 3d) the ε₂ functions (based on BSE calculations) for anatase- and rutile-TiO₂, respectively. TiO₂ is the reference material in terms of photocatalytic performances in the UV-range, while of the two phases of BiVO₄ one is a very promising candidate for photocatalysis in the visible range. In both cases, one phase (i.e. S_M-BiVO₄ and anatase-TiO₂) is known to have better photocatalytic properties than the other. Our idea is to compare these four systems and extract the main parameters that control why only two of them are efficient in the photocatalytic

activity, focusing on the optical absorption and exciton formation.

The in-plane (along x- and/or y-axis) and out-of-plane (along z-axis) contributions to the complex DF have been analyzed for the four afore mentioned materials. The in-plane and out-of-plane notation refers to the 4-fold axis in Z_T-BiVO₄, anatase and rutile, and the 2-fold axis in S_M-BiVO₄. Before discussing our predictions for BiVO₄, let us consider the two TiO₂ phases, for which experimental data are available (Figs. 3c and 3d). The insets of figures 3c and 3d provide the experimental ε₂ curves of the anatase and rutile TiO₂ phases obtained at room temperature²⁹. Very good agreement is obtained with the present BSE calculations. In particular, as previously demonstrated^{30,31}, such BSE calculations are able to reproduce the anisotropic behavior of the optical properties of anatase and rutile TiO₂. The first optical excitations are from a polarization perpendicular vs. parallel to the c-axis for anatase and rutile, respectively. In other words, anatase- and rutile-TiO₂ differ in their nature of the photo-excitations, where the former has two important polarizations (x and y) but the latter only one (z), respectively.

The optical anisotropy for S_M-BiVO₄ (Fig. 3a) is very strong in the low-energy range, in contrast to our previous observation for its related loss-function in the high-energy range. The in-plane contribution leads to a band gap of about 2.4 eV, instead of 2.9 eV for the out-of-plane contribution (see Table S1). Similarly, a strong anisotropy is found for Z_T-BiVO₄ (Fig. 3b). However, the smallest optical gap is then observed for the out-of-plane contribution (2.8 eV), while the in-plane optical gap is about 2.9 eV. In addition, the in-plane (out-of-plane) and out-of-plane (in-plane) excitations are quite intense (low) in S_M- and Z_T-BiVO₄. Such a difference between S_M- and Z_T-BiVO₄ is a direct consequence of the structural difference between these two phases, influencing the V-O and Bi-O interactions, which can easily be captured by the Bi-O-V bond angles and Bi-V distances (Table S1). Indeed, the reduction of the band gap from a pure binary oxide to the ternary compounds BiVO₄ is related to the presence of additional Bi-O and Bi-V interactions, which contribute to the anisotropy but in a different manner in the unit cell of S_M and Z_T. To be more specific, a bond angle of 90° and short Bi-V distances are preferred to have an efficient 5p(Bi)-2p(O)-3d(V) interaction. These two criteria are satisfied only for the (x,y)-plane in the S_M phase and along the z-axis, in the Z_T phase. In summary, the optical anisotropy in S_M- and Z_T-BiVO₄ is very different at low energy, leading to an intense optical excitation in the visible range for a polarization perpendicular and parallel to the c-direction for S_M- and Z_T-BiVO₄, respectively. In other words, S_M-BiVO₄ and Z_T-BiVO₄ differ in their photo-excitations, which use two polarizations (x and y) in the former but one (z) in the later compound, respectively.

We cannot conclude at this stage that such a correlation between the anisotropy of the dielectric function and the photocatalytic performances do exist, but it is interesting to note that, in both TiO₂ and BiVO₄, the better photocatalytic phase is the one, which shows an in-plane polar-

ization. Additional aspects must be considered to explain the nearly zero photocatalytic activity of $Z_T\text{-BiVO}_4$ under visible light excitation, compared to $S_M\text{-BiVO}_4$. This difference could stem from the significantly larger band gap in the Z_T -phase, which in our TB-mBJ calculations is about 2.9 eV and thus on the very edge of energies reachable by visible light.

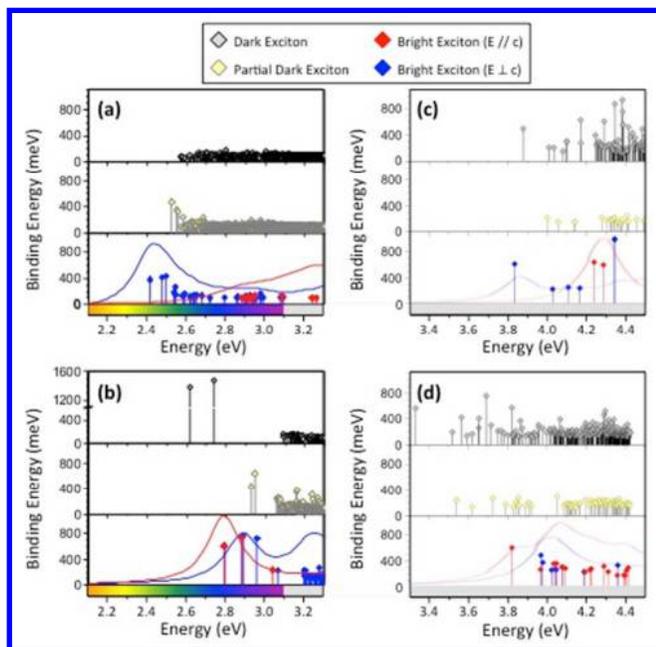


Figure 4. The first excitons calculated via the Bethe-Salpeter equation (BSE) are shown in three dark, partially dark or bright (i.e. oscillator strength is zero, small and large) for $S_M\text{-BiVO}_4$ (a), $Z_T\text{-BiVO}_4$ (b), ana- TiO_2 (c) and rut- TiO_2 (d). For each exciton the binding energy is given. For the bright excitons the imaginary part of the dielectric functions is given.

EXCITONIC EFFECTS IN BiVO_4 AND TiO_2

An exciton is by definition a correlated electron-hole, bound by a Coulombic interaction. Figure 4 shows the first thousand excitons of the four systems and their distribution in energy. Their binding energies (E_b) characterize their stability. A common feature is observed in the two photocatalytic active compounds, anatase- TiO_2 and $S_M\text{-BiVO}_4$. In both of these phases the first exciton is bright and polarized perpendicularly to the c -axis. This is in contrast to the inactive compounds rutile- TiO_2 and $Z_T\text{-BiVO}_4$, for which the first exciton is dark. In addition, the first bright exciton for rutile- TiO_2 is the 24th at 3.82 eV, with a polarization parallel to the c -axis and a strong binding energy of 605 meV.

From the present BSE calculations, we can estimate the hole-electron weighted mean distance, hereafter labeled d_{h-e} , which combined with E_b , the binding energy of the exciton, will be a measure for the probability of a hole-electron recombination, i.e. the exciton life-time. The quantity d_{h-e} has been estimated using the expression:

$$\langle d_{h-e} \rangle = \frac{\sum_{h=1}^N \sum_{c=0}^M \sum_{e=1}^N (d_{h-e}^c P_{h-e}^c)}{\sum_{h=1}^N \sum_{c=0}^M \sum_{e=1}^N P_{h-e}^c}$$

where P_{h-e}^c the probability of an hole-electron pair characterized by the distance d_{h-e}^c is given by:

$$P_{h-e}^c = \int_{r_h} d^3 r_h \int_{r_e} d^3 r_e W(r_h, r_e)$$

$W(r_h, r_e)$ the conditional probability of having electron at r_e and hole at r_h by:

$$W(r_e, r_h) = \phi(r_e, r_h) \phi^*(r_e, r_h)$$

$\phi(r_e, r_h)$ and the BSE excitonic envelop function.

The three indices (h, c and e) refer to the position of the electron (e) and the hole (h) over the N atomic sites in the cell defined by the index c, where c ranges between o (origin cell) to M. M is taken such that the maximum h-e distance is about 30 Å. The calculated probabilities for larger distances are negligible. In the present calculations the hole is systematically in the cell at the origin ($c = o$) and the bound electron in the cell c. A similar approach was used by S. Sharifzadeh *et al.*³² to estimate the average h-e distance in organic solids.

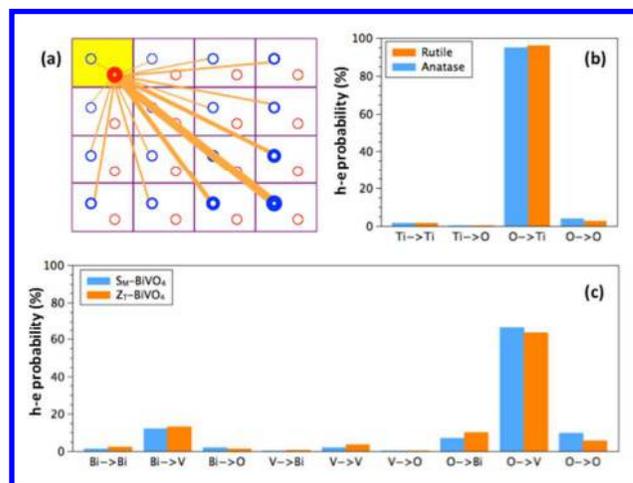


Figure 5. A schematic representation of $\langle d_{h-e} \rangle$ for a unit cell containing only two atoms (oxygen in red, metal in blue) (a), h-e probability distribution for rutile- and anatase- TiO_2 (b) and S_M - and $Z_T\text{-BiVO}_4$ (c). The photo-hole is inside the cell at the origin (defined by 000) and the photo-electron is distributed over all cells with a given h-e probability (proportional to the line width).

Figure 5a shows a schematic representation of the way $\langle d_{h-e} \rangle$ is estimated. For simplicity, the unit cell contains only two atoms. The cell at the origin is highlighted in yellow; the orange lines represent the h-e contacts. Here the hole is defined on the oxygen site inside the cell at the origin and the promoted electron on a metal site. The h-e probabilities are proportional to the width of the lines, i.e. the thicker the lines, the larger the probabilities to create an exciton between the two connected sites. By averaging all these weighted-distances, we can thus define $\langle d_{h-e} \rangle$ based on the previous formula.

Figures 5b and 5c show the distribution of the e-h probabilities for the first bright excitons of rutile- TiO_2 , anatase- TiO_2 , $S_M\text{-BiVO}_4$ and $Z_T\text{-BiVO}_4$. It clearly shows that, in both rutile- and anatase- TiO_2 , about 95% of the excitons occur between oxygen and titanium atoms (the se-

cond largest P_{e-h} is very small and corresponds to transitions between oxygen atoms due to hybridization effects). In BiVO_4 , the main contribution to the exciton spectra originates from optical transitions between oxygen and vanadium ($\text{O} \rightarrow \text{V}$, i.e. hole on O and electron on V) (66 and 63% of the transitions for $S_M\text{-BiVO}_4$ and $Z_T\text{-BiVO}_4$, respectively), much more than the bismuth to vanadium transitions (with 12 and 13% relative weight for $S_M\text{-BiVO}_4$ and $Z_T\text{-BiVO}_4$, respectively). Finally, contributions $\text{O} \rightarrow \text{Bi}$ and $\text{O} \rightarrow \text{O}$ are also observed but can be neglected. Moreover, while in TiO_2 the photo-hole was located nearly exclusively on oxygen, in BiVO_4 it is mainly positioned on oxygen but partly also on bismuth (more than 10%). Similarly, in TiO_2 the photo-electron is nearly exclusively located on the metal center (titanium). In contrast, in the vanadate ternary compounds it is distributed on vanadium (a little less than 80% with two transitions $\text{O} \rightarrow \text{V}$ and $\text{Bi} \rightarrow \text{V}$), bismuth (due to $\text{O} \rightarrow \text{Bi}$ transition, with 7 and 10%, in $S_M\text{-BiVO}_4$ and $Z_T\text{-BiVO}_4$, respectively), and oxygen (due to $\text{O} \rightarrow \text{O}$ transition, with 10 and 6%, in $S_M\text{-BiVO}_4$ and $Z_T\text{-BiVO}_4$, respectively). Such an excitonic distribution is the direct consequence of the orbital overlap in these compounds as illustrated in Figure S3 for $Z_T\text{-BiVO}_4$.

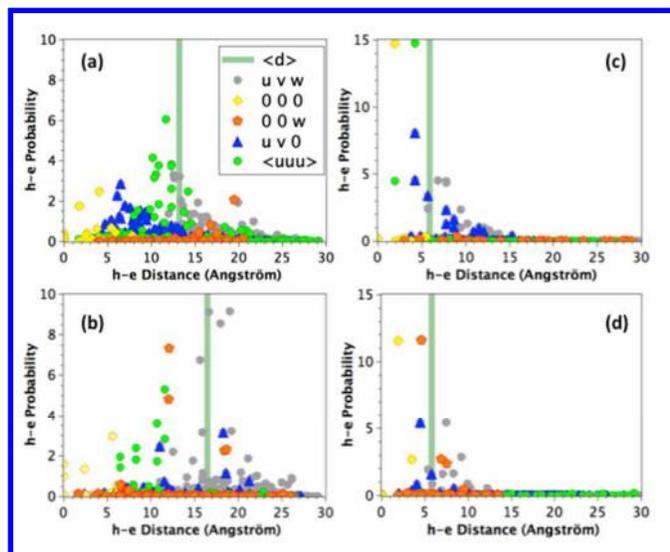


Figure 6. The h-e distance distribution of the first bright excitons for $S_M\text{-BiVO}_4$ (a), $Z_T\text{-BiVO}_4$ (b), ana- TiO_2 (c) and rut- TiO_2 (d). It corresponds to excitons #1 (at 2.41 eV / $E_b = 384$ meV), #3 (at 2.79 eV / $E_b = 610$ meV), #1 (at 3.84 eV / $E_b = 608$ meV) and #24 (at 3.82 eV / $E_b = 605$ meV), respectively for $S_M\text{-BiVO}_4$, $Z_T\text{-BiVO}_4$, anatase- TiO_2 and rutile- TiO_2 . For each compound the green bar highlights the mean h-e distance.

The optimal situation to avoid the electron-hole recombination is to have a small E_b and a large d_{h-e} . Based on the examination of average d_{h-e} and E_b values, we plot in Figure 6 the probability that an electronic transition occurs vs. its associated d_{h-e} . Only the first bright excitons of the four compounds are taken into account. Among all the photo-excitations (represented with grey circles), some specific ones are highlighted. The ones corresponding to excitons localized in the unit cell are

represented with yellow squares. The ones corresponding to an exciton that propagates in $[\bar{2}00]$, $[00\bar{2}]$ and $\langle uuu \rangle$ directions, are depicted by blue triangles, orange pentagons and green circles, respectively. Finally a green bar highlights the previously calculated average d_{h-e} distance in each material for the given exciton. Note the similarities between S_M - and $Z_T\text{-BiVO}_4$, both in terms of d_{h-e} distance and the shape of the h-e distribution. In $S_M\text{-BiVO}_4$, $Z_T\text{-BiVO}_4$ and anatase TiO_2 , the excitons which propagate along $\langle uuu \rangle$ directions have high probabilities. In contrast the highest probability in rutile TiO_2 is found for excitons propagating along the $[00w]$ direction. Figure 7 gives a very simple interpretation by showing the projection of the four structures along specific directions. Indeed, it appears that the exciton will propagate mainly in the direction of the highest packing of Bi in both BiVO_4 phases, which is along the $\langle uuu \rangle$ directions, and along the direction where the TiO_6 octahedrons are sharing their edges, which is along $\langle uuu \rangle$ and $[00\bar{2}]$ directions for anatase and rutile- TiO_2 , respectively.

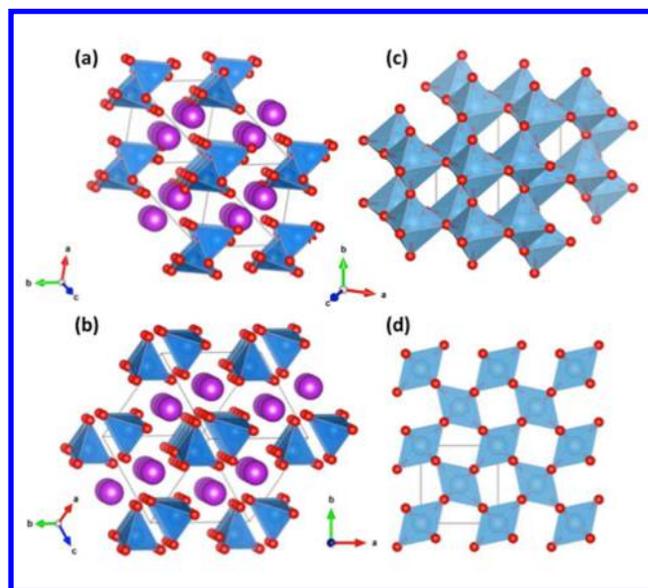


Figure 7. Projection of the atomic structure along the highest e-h probability direction for $S_M\text{-BiVO}_4$ (a), $Z_T\text{-BiVO}_4$ (b), ana- TiO_2 (c) and rut- TiO_2 (d). It corresponds to the $\langle uuu \rangle$ direction for (a), (b) and (c) and the $[001]$ direction for (d).

Another way to analyze the present BSE data is to consider the *Supplementary Tables S1-4*, which gives the excitation energy, the oscillator strength and the binding energy of the principal bright excitons in the four systems. For instance, considering the first fifty excitons, 13 are bright in $S_M\text{-BiVO}_4$, but only 5 in $Z_T\text{-BiVO}_4$. Similarly, among the first fifties excitons, 10 are bright in anatase- TiO_2 , while only 1 in rutile TiO_2 . In summary, under visible light excitation, more electron-hole pairs are formed in S_M - than in $Z_T\text{-BiVO}_4$, and these excitons have the propensity to recombine more frequently in Z_T - than in $S_M\text{-BiVO}_4$. A similar analysis could be done for anatase- and rutile- TiO_2 .

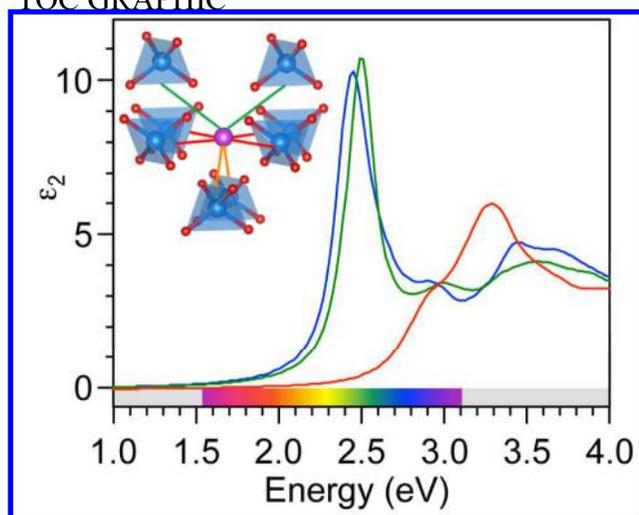
CONCLUSION

In conclusion, the present investigation evidences many common features between the reference UV-light photocatalytic compound anatase-TiO₂ and the promising candidate for visible-light photocatalytic compound S_M-BiVO₄. Indeed, both exhibit an indirect band gap, their first exciton is bright and shows a polarization perpendicular to the *c*-axis. The fact that the band gap of the two photocatalytic active materials is indirect is essential in the sense that it prevents a direct recombination of the hole-electron pair, leading to longer lifetime of photoexcited electrons and holes than direct band gap rutile-TiO₂ and Z_T-BiVO₄. In addition, the first excitons of these two last compounds are dark and show a polarization parallel to the *c*-axis. These inter-related properties appear to be essential and should be considered with care in the quest of an optimal photocatalytic material.

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Author Contributions

XR and SJ initiated the project, LL performed the EELS measurements and their analyses, TD, PB and XR performed the DFT simulations, RL performed the BSE calculations. All authors analyzed the data and wrote the paper and contributed to corrections and modifications.

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