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Delicate Balance Among Three Ferroic Polymorphic Phases in $\text{BiFeO}_3\text{-BiMnO}_3\text{-PbTiO}_3$

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ABSTRACT: Phase-change functional responses can be obtained in perovskite solid solutions at structural instabilities involving multiferroic states, a very promising approach for realizing magnetoelectric effects at room temperature. A line of multiferroic morphotropic phase boundaries (MPBs) has been described within the phase diagram of the $\text{BiFeO}_3\text{-BiMnO}_3\text{-PbTiO}_3$ ternary system, in which an evolution from $Cc/P4mm$ -type (polar/polar) to $Pnma/P4mm$ -type (antipolar/polar) MPBs takes place. Coexistence of the three polymorphs results in a small region, where distinctive phase-change magnetoelectric responses are anticipated. In this work, all three polymorphs were obtained, either isolated or in phase coexistence with varying percentages, by tailoring the cooling profile after high temperature annealing in a small compositional range. Perovskite structure was characterized, and temperature dependent magnetic and electrical characterizations were carried out to define the ferroic orders and crystal physicochemical properties of the three polymorphs. Results revealed the close interplay among the crystal structure, point defects, and multiferroism in this small region, where the delicate balance among the three polymorphs is shown. Additionally, the effect of the size reduction across the submicron range, down to the nanoscale, is described and discussed.

1. INTRODUCTION

Multiferroism refers to materials in which two or more types of primary ferroic ordering are present in the same phase, being of particular interest those with coexistence of the electrical and magnetic orders.¹⁻³ These materials develop a linear electrical polarization (magnetization) in response to an applied magnetic (electric) field, and are also liable to show non-linear coupling phenomena like crossed magnetoelectric switching.⁴ This could enable a novel nonvolatile random access memory, that is, a low-power, high-density information storage that would combine fast electrical WRITE/magnetic READ operations, among a number of new potentially disruptive technologies like voltage-controlled spintronic devices, uncooled magnetic sensors, or electrically-tunable magnetic microwave components.⁵⁻⁷

A major research activity is concentrated on single-phase multiferroics, particularly on those that might exhibit such coupling at room temperature, among which ABO_3 perovskite oxides stand out. This is because multiferroicity can be chemically engineered on a site-by-site basis, by placing ferroelectrically and magnetically active cations in *A*- and *B*-site, respectively, so that the two ferroic orders develop independently.⁸ This is the case of BiFeO_3 and BiMnO_3 ,^{9,10} two systems at the cutting edge of research, particularly in the context of strained films,^{11,12} and of the solid solutions they form with PbTiO_3 .^{13,14} Interest in solid solutions exhibiting coexistence of multiferroic polymorphs at the morphotropic phase boundary (MPB) is increasing, not only for property enhancement takes place but also for phase-change functional responses can be obtained.^{15,16}

Strong phase-change magnetoelectric responses were first anticipated in the perovskite $\text{BiFeO}_3\text{-BiCoO}_3$ binary system by a first-principles investigation, associated with the coexistence of multiferroic phases of rhombohedral $R3c$ and tetragonal

$P4mm$ symmetries at the MPB.¹⁷ Mechanism proposed was the discrete rotation of magnetization following that of polarization under the electric field, associated with an electric field induced transformation between the two multiferroic polymorphs.¹⁸ Phase-change magnetoelectric phenomena have also been found in other perovskites like in Nd substituted BiFeO_3 , associated this time to an electric field-induced orthorhombic $Pnma$ to rhombohedral $R3c$ phase transition.¹⁹ Pressure-induced transformation between multiferroic polymorphs has also been reported, like in La substituted BiFeO_3 .²⁰ In this case, a transition from rhombohedral/polar $R3c$ to orthorhombic/antipolar $Pbam$ phases was found under high pressure, and strong effects on properties were anticipated. Therefore, phase-change functional responses might be a general property of multiferroic MPBs.²¹

We have recently reported the occurrence of phase-change magnetoelectric phenomena for a specific composition of the $\text{BiFeO}_3\text{-BiMnO}_3\text{-PbTiO}_3$ ternary system, for which coexistence of phases with monoclinic Cc and tetragonal $P4mm$ symmetries was found.²² In this material, both polymorphs are ferroelectric and antiferromagnetic (AFM), yet a weak ferromagnetic component appears within the Cc phase, as a result of a spin reorientation (SRO) transition within its AFM state.²³ Room-temperature magnetoelectricity is obtained, not only due to an electric field driven phase-change between the two multiferroic polymorphs, but also by the presence of a transverse lattice softening within the monoclinic phase that gives rise to linear magnetoelectric effect.²² The latter phenomenology would be analogous to that described in high-sensitivity MPB ferroelectrics, which is responsible for the very-high piezoelectric responses.²⁴ Indeed, a theoretical study has pointed out to the possibility of obtaining enhanced magnetoelectric responses at multiferroic analogues associated with this lattice transverse softening.²⁵

A close interplay among the crystal structure, the defect chemistry and multiferroic properties of phases has been recently demonstrated for compositions of this ternary system, yet at a different MPB.²⁶ Namely, phase coexistence of a ferroelectric $P4mm$ with an antiferroelectric $Pnma$ polymorph was found in $\text{Bi}_{0.4}\text{Pb}_{0.6}\text{Mn}_{0.4}\text{Ti}_{0.6}\text{O}_3$, where both showed spin-glass-like magnetic behavior. This type of polar/antipolar MPB has also been proposed to engineer new high-sensitivity ferro-/piezoelectrics.²⁷ Besides, large phase-change magnetoelectric responses have been anticipated as a result of an antipolar to polar transition.¹⁹ Point defects are thought to be behind changes in the crystal structure reported by different groups with the processing conditions used, as well as with thermal history.^{28–30}

This work provides a comprehensive analysis of the delicate balance between polymorphs along the line of MPBs present in the BiFeO_3 - BiMnO_3 - PbTiO_3 ternary system,³¹ its relation to the defect chemistry, and its repercussion in the multiferroic properties. Moreover, a grain size effect, not described before, is revealed and discussed accordingly. Materials chosen are compositionally located in the region where the evolution from $Cc/P4mm$ -type to $Pnma/P4mm$ -type MPBs takes place, at which coexistence of up to three different polymorphs can be obtained. These could be an alternative to BiFeO_3 - BiCoO_3 for room-temperature phase-change magnetoelectric effect, with the advantage that they can be easily obtained by mechanochemical methods. The high chemical homogeneity provided by this synthesis technique has been key to uncover the MPB phases, and more importantly, to establish their multiferroic features. High-resolution x-ray diffraction (XRD) has allowed assessing the symmetry of the phases involved. Results of the dielectric and magnetic characterizations have allowed establishing the structure-property relationships, and to discuss the key role of the defect chemistry on the polymorph stabilized at room temperature.

2. EXPERIMENTAL SECTION

Nanocrystalline powders with perovskite structure were obtained by mechanosynthesis in a high energy planetary mill (Fritsch Pulverisette 6) operating at 300 rpm for 24 h in tungsten carbide milling media, from stoichiometric mixtures of analytical grade oxides Bi_2O_3 (Aldrich, 99.9% pure), PbO (Merck, 99%), Fe_2O_3 (Cerac, 99.9%), TiO_2 (anatase, Cerac, 99%) and Mn_2O_3 (Aldrich, 99%). Details of the procedure and of the mechanisms that take place during mechanical treatments can be found elsewhere.³¹ The milling conditions were tailored so as negligible contamination levels resulted. Compositions prepared respond to the general formula $\text{Bi}_{x+y}\text{Pb}_{1-x-y}\text{Fe}_x\text{Mn}_y\text{Ti}_{1-x-y}\text{O}_3$, with $x = 0.52$; $y = 0.1$ (labeled 1), $x = 0.47$; $y = 0.15$ (2), $x = 0.47$; $y = 0.11$ (3), and $x = 0.7$; $y = 0.05$ (4). Their actual locations in the phase diagram of the ternary system are indicated in Fig. 1. Mechanosynthesized powders were thermally treated at 950 °C for 12 h to increase crystal sizes for the structural characterization. Two cooling profiles were used: (i) a slow cooling with rate of 3 °C min⁻¹ and (ii) a fast cooling (quenching) in air down to room temperature.

Crystal structure of powdered samples was assessed by XRD. A highly accurate two-axis diffractometer with a Rigaku RAHF18 rotating anode generator, in Bragg-Brentano geometry and coupled with a furnace, was used in the range of 15° to 150°

(2θ), in steps of 0.02° and counting time of 20 s per step. $\text{CuK}\alpha$ doublet ($\lambda = 1.5418$ Å) was used in the experiments, and a monochromator producing $\text{CuK}\beta$ ($\lambda = 1.39225$ Å) was occasionally used for higher resolution. Diffraction profiles were analyzed by the Rietveld method implemented in Jana2006 package.³²

Magnetic properties were characterized with a Quantum Design MPMS-XL5 SQUID magnetometer. Powdered samples were cooled down to 2 K under zero field and then a magnetic field of 500 Oe was applied and maintained during a heating/cooling (ZFC/FC) cycle up to 400 K.

Ceramic materials for the dielectric characterization were prepared by direct hot-pressing of the nanocrystalline powders. Temperature was varied between 1000 and 800 °C in order to obtain a range of microstructures with decreasing grain size. A holding time of 1 h and a pressure of 60 MPa were selected, while heating/cooling rates of 3 °C min⁻¹ were used. Archimedes' method was used to measure the density of the sintered pellets, and final densification levels above 99% were systematically attained.

Ceramic microstructure was characterized by field-emission scanning electron microscopy with a NovaTM NanoSEM 230 apparatus. Samples were polished to mirror finish with Al_2O_3 suspensions down to 0.1 μm, and thermally etched at 800 °C for 20 min and quenched in air down to room temperature to reveal the grain boundaries. A series of very dense microstructures with tailored average grain size across the submicron range, down to the nanoscale was obtained, as shown in Fig. 2 for ceramics of composition 1, taken by way of example. Note the absence of residual porosity or secondary phases. The evolution of the average grain size with hot-pressing temperature suggests a single mechanism of grain growth [Fig. 2(d)].

Figure 1. Schematic phase diagram of the BiFeO_3 - BiMnO_3 - PbTiO_3 ternary system, where MPB regions are highlighted and nominal compositions under study indicated with closed circles.

Figure 2. Scanning electron microscopy micrographs of ceramics of composition 1 ($\text{Bi}_{0.62}\text{Pb}_{0.38}\text{Fe}_{0.52}\text{Mn}_{0.1}\text{Ti}_{0.38}\text{O}_3$) obtained by hot-pressing at (a) 850 °C, (b) 900 °C and (c) 950 °C, along with the resulting size distributions in each case. (d) Evolution of the average grain size with the sintering temperature (standard deviations, SD, are given as error bars). Grain sizes (μm) are given in the inserted table.

Electrical properties were measured on ceramic discs, on which Ag electrodes were painted and annealed at 750 °C for 1 h. Dependences of the dielectric permittivity and losses on temperature were dynamically measured between room temperature and 700 °C under heating/cooling cycles at 1.5 °C min^{-1} with a HP4284A precision LCR Meter at several frequencies between 100 Hz - 1 MHz.

3. RESULTS AND DISCUSSION

3.1. Structural Characterization. Figure 3 shows the XRD patterns of mechanosynthesized powders corresponding to the MPB compositions displayed in Fig. 1, after thermal treatment at 950 °C / 12 h followed by two different cooling profiles, that is, slow cooling (black patterns) and quenching (red patterns). In the composition 1 ($\text{Bi}_{0.62}\text{Pb}_{0.38}\text{Fe}_{0.52}\text{Mn}_{0.1}\text{Ti}_{0.38}\text{O}_3$), both powders crystallized into perovskite-type structure (no secondary phase was found), and peak splitting clearly evidences that phase coexistence between polymorphs of different symmetries occurs [Fig. 3(a)]. In the quenched material, this coexistence involves a tetragonal polymorph and a second one that might be either rhombohedral or monoclinic, whereas the slowly cooled sample showed broad/asymmetric diffraction peaks that also indicate phase coexistence, though actual symmetries, in this case, cannot be directly assessed.

Composition 2 ($\text{Bi}_{0.62}\text{Pb}_{0.38}\text{Fe}_{0.47}\text{Mn}_{0.15}\text{Ti}_{0.38}\text{O}_3$) also showed the powders crystallized into perovskite-type structure [Fig. 3(b)], yet a clear phase coexistence was only evidenced in the quenched material, which also seems to involve tetragonal and rhombohedral (or monoclinic) polymorphs, whereas a single polymorph that appears to be cubic (or likely pseudo-cubic) was obtained after slow cooling. On the contrary, composition 3 ($\text{Bi}_{0.58}\text{Pb}_{0.42}\text{Fe}_{0.47}\text{Mn}_{0.11}\text{Ti}_{0.42}\text{O}_3$) seems to be fully tetragonal after quenching, whereas the slowly cooled material showed broad/asymmetric diffraction peaks [Fig. 3(c)], like those for composition 1. Results thus confirm these three compositions are at or nearby the line of MPBs in the ternary system, and show that different polymorphs can be stabilized in the same material depending on the cooling profile used.

Diffraction profiles were analyzed by the Rietveld method to establish the actual crystal symmetry of the coexisting polymorphs, for which tetragonal $P4mm$ space group of PbTiO_3

along with the rhombohedral $R3c$ of BiFeO_3 or monoclinic Cc symmetries (reported for BiFeO_3 - PbTiO_3 system)²⁸ were chosen as structural models in the refinements of the quenched samples. However, differences found on the statistical parameters resulting from models involving either $R3c$ or Cc were not large enough to fully discriminate which of these two symmetries coexists with the $P4mm$ one. This is due to the complexity of the profiles resulting from the phase coexistence, a handicap for the successful application of the Rietveld method to MPB compositions, further complicated by the crystal stresses present due to the difference in transition strain between the polymorphs and anisotropy factors due to ferroelectricity.

The study was then carried out in a composition with nominal formula $\text{Bi}_{0.75}\text{Pb}_{0.25}\text{Fe}_{0.7}\text{Mn}_{0.05}\text{Ti}_{0.25}\text{O}_3$ (labeled 4), which is located well within the pseudo-rhombohedral region of the phase diagram, and away from the line of MPBs. As expected, only one polymorph is present in the patterns of samples either slowly cooled or quenched, as shown in Fig. 4(a) for the slowly cooled one. The absence of the tetragonal phase, in this case, avoids the complexity of refining patterns with phase coexistence. The profile was refined for both $R3c$ and Cc symmetries, among others, and results are given in Table 1. It can be seen that statistical parameters are slightly better for the Cc space group, in particular GOF indicator and R_{Bragg} factor.

The best agreement for the Cc phase is also evident when one specifically analyses the refined profiles across the diffraction peaks, such as $(112)_p$ and $(103)_p$ ones (referring to pseudocubic indices), as illustrated in Figs. 4(b) and 4(c). It is clear that the fit using $R3c$ symmetry is not satisfactory, as the anomalous peak broadening is not consistent with considering them as doublets, as expected for this symmetry. This confirms that the polymorph in coexistence with the tetragonal one is most likely the monoclinic Cc in these BiFeO_3 -rich compositions, so that a $Cc/P4mm$ -type MPB would be obtained. It should be taken into account the greater number of structural degrees of freedom that the Cc structure introduces in the refinement, as compared to the $R3c$ one. Indeed, a controversy persists regarding the actual existence of the Cc polymorph in the BiFeO_3 - PbTiO_3 binary system.^{10,22,28} However, the presence of transverse lattice softening and a magnetic SRO transition,^{14,23} both phenomena characteristic of the monoclinic symmetry, seems to confirm the presence of the Cc phase in the ternary system.

chosen as structural models in the refinements. The best fit between the observed and calculated profiles was obtained for

Figure 3. XRD patterns with $\text{CuK}\alpha$ at room temperature on powder samples thermally treated at 950°C and using different cooling rates: slow cooling (black pattern) and quenching (red pattern), corresponding to compositions (a) 1 - $\text{Bi}_{0.62}\text{Pb}_{0.38}\text{Fe}_{0.52}\text{Mn}_{0.1}\text{Ti}_{0.38}\text{O}_3$, (b) 2 - $\text{Bi}_{0.62}\text{Pb}_{0.38}\text{Fe}_{0.47}\text{Mn}_{0.15}\text{Ti}_{0.38}\text{O}_3$, (c) 3 - $\text{Bi}_{0.58}\text{Pb}_{0.42}\text{Fe}_{0.47}\text{Mn}_{0.11}\text{Ti}_{0.42}\text{O}_3$. Miller indices are for pseudo-cubic symmetry.

Additionally, the symmetry of the pseudo-cubic polymorph that resulted from slow cooling of composition 2 was also investigated. In this case, orthorhombic *Pnma* space group, among others such as cubic *Pm-3m* and monoclinic *C2/c* or *C2* (typically reported for BiMnO_3 -based compositions),³³ were

Figure 4. High-resolution XRD pattern using $\text{CuK}\alpha$ on powder sample of composition 4 ($\text{Bi}_{0.75}\text{Pb}_{0.25}\text{Fe}_{0.7}\text{Mn}_{0.05}\text{Ti}_{0.25}\text{O}_3$) along with the Rietveld refinement using monoclinic Cc symmetry. Y_{Obs} and Y_{Cal} refer to measured and calculated intensities, respectively. Rietveld refinements with Cc and rhombohedral $R3c$ symmetries across the (b) $(112)_p$ and (c) $(103)_p$ reflections (pseudo-cubic indices) are shown to illustrate the better agreement of Cc phase.

the $Pnma$ symmetry, as shown in Fig. 5(a), though differences with the cubic $Pm-3m$ instead were not that large ($\text{GOF} = 1.39$ for $Pnma$ and 1.43 for $Pm-3m$). On the other hand, the quenched material showed the best fit from the model involving coexistence of Cc (92%) and $P4mm$ (8%) phases, as shown in Fig. 5(b). The refined structural and statistical parameters of composition 2 are given in Table 2.

In order to deepen into the symmetry of this apparently pseudocubic polymorph in the slowly cooled sample of composition 2, the temperature evolution of the perovskite phase was investigated. Figure 6 shows the XRD profiles around the $(110)_p$ peak (referring to pseudocubic index) at increasing temperatures, along with the resulting temperature dependence of the lattice parameter assuming to be pseudocubic (a_{ps}). Two regions are clearly differentiated with a distinctive change of the thermal expansion rate at $\sim 500^\circ\text{C}$. A similar change of slope was reported for $\text{Bi}_{0.4}\text{Pb}_{0.6}\text{Mn}_{0.4}\text{Ti}_{0.6}\text{O}_3$ (with a $Pnma/P4mm$ -type MPB), and associated with the transition from the orthorhombic $Pnma$ towards the high-temperature parent phase,²⁶ which turns out to be cubic $Pm-3m$ in most

Table 1. Results of the Rietveld refinements of $\text{Bi}_{0.75}\text{Pb}_{0.25}\text{Fe}_{0.7}\text{Mn}_{0.05}\text{Ti}_{0.25}\text{O}_3$ sample (composition 4).^a

Space group	<i>R3c</i>	<i>Cc</i>
Statistical parameters	$R_{\text{wp}} = 12.12$; $R_{\text{p}} = 8.97$; $R_{\text{exp}} = 9.18$;	$R_{\text{wp}} = 11.19$; $R_{\text{p}} = 8.41$; $R_{\text{exp}} = 9.17$;
	$R_{\text{Bragg}} = 6.06$; GOF = 1.32	$R_{\text{Bragg}} = 4.25$; GOF = 1.22
Lattice parameters (Å)	$a = 5.586(5)$; $c = 13.842(9)$	$a = 9.763(4)$; $b = 5.589(3)$; $c = 5.634(7)$; $\beta = 125.85(8)^{\circ}$

Space group	<i>Cc</i>				
	Bi/Pb	Fe/Mn/Ti	O1	O2	O3
<i>x</i>	0	0.275(1)	0.044(2)	0.327(5)	0.283(0)
<i>y</i>	0.25	0.243(1)	0.291(7)	0.453(6)	-0.027(5)
<i>Z</i>	0	0.731(0)	0.460(2)	0.033(7)	-0.049(1)
<i>B</i> _{eq} (Å ²)	0.93	0.01	1.50	1.50	1.50

^aReliability factors (R -factors) and goodness-of-fit indicator (GOF) are given for both rhombohedral $R3c$ and monoclinic Cc symmetries, while position coordinates (x, y, z) and thermal parameters (B_{eq}) are given for the Cc symmetry.

ferroelectric perovskites.³⁴ Results, therefore, indicate the symmetry of this polymorph to be $Pnma$ at room temperature.

The strong dependence of the phase coexistence on the thermal history in these MPB compositions is worth further discussion. A major role of the defect chemistry can be anticipated, in particular, of the oxygen vacancies generated by the different cooling profiles, and the resulting partial reduction/oxidation of the transition metal cations.^{35,36} This means that the equilibrium

oxygen content during the thermal treatment might be frozen at room temperature by quenching, but a slow cooling provides more time for oxygen equilibrium, thus involving oxidation/reduction of cations like Fe and Mn. This would imply a change in the tolerance factor that might push the system across the MPB. The point defect equilibrium here can be expressed as:

where M_M^* refers to $(\text{Fe/Mn})^{4+}$ and M_M to $(\text{Fe/Mn})^{3+}$, V_O^* is an oxygen vacancy and O_O is an oxygen anion.

Note the lattice expansion that resulted from the quenching treatment, so as the unit cell volume of the stabilized polymorph is about 0.33% and 2% higher than that obtained by slow cooling, in the compositions 2 and 3, respectively. This favored the formation of the *Cc* phase in the former and the *P4mm* one in the latter. This chemical expansion can be understood from the difference in the ionic radii of the cations involved, and the resulting changes in the size of the MO_6 octahedra.³⁵ The smaller unit cell volume in both materials after the slow cooling suggests that chemical contraction due to Fe and Mn oxidation took place, whereas a frozen oxygen equilibrium state, achieved with the quenching, must hinder cations oxidation.²⁶

It is worth summarizing the results of the structural characterizations before addressing the ferroic properties. Coexistence of the monoclinic *Cc* and tetragonal *P4mm* polymorphs has been shown for compositions 1 and 3, while coexistence of all three *Cc*, *P4mm* and *Pnma* has been obtained for composition 2. Phase percentages can be varied by high-temperature treatments with tailored cooling profiles, so that the different polymorphs have been mostly isolated at room temperature: monoclinic *Cc* in powders of compositions 1 and 3 slowly cooled, orthorhombic *Pnma* in composition 2 slowly cooled, and tetragonal *P4mm* for composition 3 after quenching. This ensemble of samples with close compositions and tailored perovskite structure locates in a small region of the phase diagram around the line of MPBs, where the evolution from *Cc*/*P4mm*-type to *Pnma*/*P4mm*-type takes place.

Figure 5. High-resolution XRD patterns using $\text{CuK}\beta$ on powder samples of composition 2 ($\text{Bi}_{0.62}\text{Pb}_{0.38}\text{Fe}_{0.47}\text{Mn}_{0.15}\text{Ti}_{0.38}\text{O}_3$), after thermal treatment at 950 °C followed by (a) slow cooling and (b) quenching. Profiles were analyzed by the Rietveld method using (a) orthorhombic *Pnma* and (b) coexistence of monoclinic *Cc* and tetragonal *P4mm* symmetries. Y_{Obs} and Y_{Cal} refer to the measured and calculated intensities, respectively.

Table 2. Results of the Rietveld refinements of $\text{Bi}_{0.62}\text{Pb}_{0.38}\text{Fe}_{0.47}\text{Mn}_{0.15}\text{Ti}_{0.38}\text{O}_3$ sample (composition 2), after (a) slow cooling and (b) quenching, using *Pnma* in the former, and coexistence of *Cc* and *P4mm* in the latter.^a

(a)		<i>Pnma</i>			
		Bi/Pb	Fe/Mn/Ti	O1	O2
<i>x</i>		0.006(7)	0	0.507(6)	0.238(5)
<i>y</i>		0.25	0	0.25	0.003(6)
<i>z</i>		0.002(4)	0.5	0.033(3)	0.756(5)
<i>B</i> _{eq} (Å ²)		2.66	0.67	1.50	1.50
Lattice parameters (Å)		<i>a</i> = 5.602(0); <i>b</i> = 7.913(4); <i>c</i> = 5.601(1)			
Statistical parameters		<i>R</i> _{wp} = 9.24; <i>R</i> _P = 8.11; <i>R</i> _{exp} = 6.65; <i>R</i> _{Bragg} = 9.62; GOF = 1.39			

(b)		<i>Cc</i>				<i>P4mm</i>				
		Bi/Pb	Fe/Mn/Ti	O1	O2	O3	Bi/Pb	Fe/Mn/Ti	O1	O2
<i>x</i>		0	0.271(0)	-0.015(2)	0.335(9)	0.300(6)	0	0.5	0.5	0.5
<i>y</i>		0.25	0.246(0)	0.25	0.456(6)	0.037(9)	0	0.5	0	0.5
<i>z</i>		0	0.726(0)	0.559(7)	0.060(0)	-0.040(0)	0	0.518(8)	0.565(0)	0.053(1)
<i>B</i> _{eq} (Å ²)		1.88	0.79	1.50	1.50	1.50	1.88	1.58	1.50	1.50
Lattice parameters (Å)		<i>a</i> = 9.773(7); <i>b</i> = 5.583(1); <i>c</i> = 5.615(4); <i>β</i> = 125.60(7)°				<i>a</i> = 3.873(3); <i>c</i> = 4.236(4)				

Statistical parameters

 $R_{wp} = 10.87$; $R_p = 9.38$; $R_{exp} = 6.75$; $R_{Bragg}(Cc) = 15.65$; $R_{Bragg}(P4mm) = 31.08$; $GOF = 1.61$

Phase fraction

0.92 (*Cc*) / 0.08 (*P4mm*)^aPosition coordinates (*x*, *y*, *z*), thermal parameters (*B*_{eq}), reliability factors (*R*-factors) and goodness-of-fit indicator (*GOF*) are given.

Figure 6. Temperature evolution of XRD profiles using $\text{CuK}\beta$ around the $(110)_p$ reflection (pseudocubic index) for powder sample of composition 2 ($\text{Bi}_{0.62}\text{Pb}_{0.38}\text{Fe}_{0.47}\text{Mn}_{0.15}\text{Ti}_{0.38}\text{O}_3$) after slow cooling. (b) Temperature evolution of the lattice parameter obtained from Le Bail analysis assuming to be pseudo-cubic (a_{ps}). Colored regions indicate different polymorphs.

3.2. Magnetic Properties. Figure 7 shows the temperature dependence of the molar magnetic susceptibility (M/H ratio) of powdered samples corresponding to compositions under study, after either slow cooling or quenching treatments, to establish the magnetic features of the different polymorphs. Let's first analyze the results for composition 4, which was structurally identified as monoclinic *Cc*, and whose magnetic behavior indicates a weak spin-canted ferromagnetic component below 400 K [Fig. 7(a)]. No differences were found between slowly cooled and quenched samples in this case. The behavior is very similar to that reported for compositions at the MPB of the BiFeO_3 - PbTiO_3 binary system,³¹ for which the anomaly was associated with a spin reorientation phenomenon within the AFM state.²³ Indeed, a Néel temperature above 400 K has been reported for the 0.7BiFeO_3 - 0.3PbTiO_3 (at 480 K),³⁷ a composi-

tion close to that labeled 4 in this work. Below the SRO transition, T_{SRO} , the ferromagnetic component undergoes a spin flop, so that a larger net magnetic moment results.²² A weak ferromagnetic component would be highly desired to obtain a linear magnetoelectric effect, and indeed, it could be very useful in the search of room-temperature responses.

A similar magnetic behavior was obtained in the samples of compositions 1 and 2 after slow cooling and quenching treatments, respectively, in which the predominant polymorph is again the monoclinic *Cc* one. ZFC/FC curves again showed the SRO phenomenon and spin-canting ferromagnetism [Figs. 7(b) and 7(c)]. Note the shift of the SRO transition towards lower temperatures upon addition of BiMnO_3 (that is, $T_{\text{SRO}} = 222$ and 168 K for the compositions 1 and 2, respectively). Note also the small irreversibility of the ZFC/FC curves above the SRO transition that indicates the system to still present AFM ordering above it, and also beyond the measuring range. This seems to be the general picture of the magnetic behavior of monoclinic *Cc* phase within the ternary system under study.

A different behavior was found for the quenched material of composition 3, in which the polymorph that largely prevails at room temperature is the tetragonal *P4mm* one. ZFC/FC curves in this case showed mainly an overall near-paramagnetic behavior down to very low temperatures [Fig. 7(d)]. It should be noted that the small anomaly at ~ 160 K, along with the tiny irreversibility in the whole measuring range, are both rather related with the presence of a very small percentage of monoclinic phase than with the major tetragonal one. Indeed, we have recently shown that the anomaly corresponding to the AFM ordering of the tetragonal polymorph takes place at about 200 K for the BiFeO_3 - PbTiO_3 binary system,¹⁶ but immediately shifted below 100 K upon adding a small amount of BiMnO_3 .²² Therefore, a further shift to very low temperatures (see the small divergence about 2 K), or even a vanishes, of the tetragonal AFM transition could be anticipated for the MPB compositions under study.

The magnetic behavior of these MPB compositions can then significantly differ depending on whether the material was quenched or slowly cooled from high temperature, a phenomenon that is closely related with the polymorph that results from the thermal treatment. This is the case of composition 2, for which ZFC/FC curves in the slowly cooled sample clearly indicated overall near-paramagnetic behavior down to very low temperatures [Fig. 7(e)], a material that was structurally identified as orthorhombic *Pnma*. Again, the small anomaly at ~ 165 K and tiny irreversibility seem both to be related with the presence of traces of monoclinic phase. Indeed, this was the predominant magnetic feature of the quenched material [Fig. 7(c)].

Figure 7. Zero-field-cooled/Field-cooled (ZFC/FC) susceptibility curves under a *dc* magnetic field of 100 Oe for powder samples of composition (a) 4 - $\text{Bi}_{0.75}\text{Pb}_{0.25}\text{Fe}_{0.7}\text{Mn}_{0.05}\text{Ti}_{0.25}\text{O}_3$, (b) 1 - $\text{Bi}_{0.62}\text{Pb}_{0.38}\text{Fe}_{0.52}\text{Mn}_{0.1}\text{Ti}_{0.38}\text{O}_3$ slowly cooled, (c) 2 - $\text{Bi}_{0.62}\text{Pb}_{0.38}\text{Fe}_{0.47}\text{Mn}_{0.15}\text{Ti}_{0.38}\text{O}_3$ with quenching, (d) 3 - $\text{Bi}_{0.58}\text{Pb}_{0.42}\text{Fe}_{0.47}\text{Mn}_{0.11}\text{Ti}_{0.42}\text{O}_3$ with quenching and (e) 2 slowly cooled. Crystal structure is given for each case. T_{SRO} indicates the magnetic anomaly associated with spin-reorientation phenomenon within the monoclinic *Cc* phase.

Results indicate that only the monoclinic phase presents AFM ordering and SRO phenomena close to room temperature, while magnetism vanishes for the tetragonal and orthorhombic polymorphs. This coexistence of polymorphs in the same material with a differentiated magnetism could be very useful for obtaining phase-change magnetoelectric responses.

The origin of this distinct magnetism among polymorphs deserves further discussion. On the one hand, both the AFM ordering and SRO phenomena associated to the monoclinic phase seem to be related with superexchange pathways on the Fe^{3+} sublattice. Addition of BiMnO_3 increases chemical disorder, and gives rise to competing interactions between both the Fe and Mn magnetic species. This would explain the shift to lower temperature of the SRO anomaly of the monoclinic phase upon

increasing BiMnO_3 . Moreover, defect chemistry and order/disorder features of this complex system need to be also considered. The role of manganese, which is known to readily adopt different oxidation states in perovskites, *e.g.* Mn^{3+} and Mn^{4+} , after the formation of cation vacancies that are charge compensated by the mixed-valence state.³⁶ Indeed, *Pnma* phase of the BiMnO_3 - PbTiO_3 binary system, and its magnetic spin-glass behavior, have been associated with the presence of $\text{Mn}^{4+}/\text{Mn}^{3+}$ and the resulting competing interactions.²⁶ The strong competition between superexchange of ferromagnetic ($\text{Mn}^{4+}\text{-O-Mn}^{3+}$) and negative AFM ($\text{Mn}^{4+}\text{-O-Mn}^{4+}$ and $\text{Mn}^{3+}\text{-O-Mn}^{3+}$) interactions lead to freezing of the local magnetic moment and spin-clustering with random orientations.³⁸ This would explain the

absence of long-range order in both $P4mm$ and $Pnma$ phases within this ternary system.

3.3. Dielectric Properties. Figure 8 shows the temperature dependences of the relative permittivity (K') and dielectric losses ($\tan \delta$) of ceramics processed at 950 °C, corresponding to the different compositions under study. Dielectric curves at several frequencies during heating stage are given in Fig. 8(a) for sample of composition 1, taken by way of example. Similar frequency dependence of the dielectric behavior was found for all compositions. Note the Maxwell-Wagner relaxation below 400 °C, that is, a step-like increase in K' at a temperature that shifts with frequency, a phenomenon usually associated with the presence of significant electrical conductivity in these materials.³⁹ This is also evident in $\tan \delta$ curves, directly proportional to conductivity, in which large frequency dispersion and high losses were obtained. The anomaly associated with the transition from the ferroelectric to paraelectric phase is observed at $T_c = 575$ °C for this composition in ceramic form, which is expected to be monoclinic Cc . A cooling rate of 3 °C min⁻¹ was used during the hot pressing, so room temperature phases should be those described for slowly cooled powders. XRD patterns of ceramics are consistent with this hypothesis, though they showed an increased broadening associated with the lower crystallinity and the ceramic stress field that hinders phase determination.

Figure 8(b) shows the dielectric curves during a heating/cooling cycle at 1 MHz for compositions 1, 2 and 4 (high-frequency data is given to minimize conduction contributions). The dielectric anomaly associated with the ferroelectric transition shifted to lower temperatures, in both permittivity and losses, with increasing the amount of Mn in these materials (that is, 4 → 1 → 2), along with a decrease of maximum permittivity, which is consistent with previous works.³¹ All dielectric curves showed the typical hysteresis on a heating/cooling cycle for a first-order ferroic transition, like those obtained in BiFeO₃-rich compositions of the BiFeO₃-PbTiO₃ binary system.¹⁴ Samples of compositions 4 and 1 were monoclinic Cc after a slow cooling, and thus, a ferroelectric transition takes place in this case. However, powders of composition 2 were clearly $Pnma$ and this polymorph was recently claimed as antiferroelectric.²⁶ The presence of a dielectric anomaly supports this assertion and thus, a transition from antiferroelectric to paraelectric seems to take place for this specific material.

Figure 8. Temperature dependences of the relative permittivity (K') and (inset) dielectric losses ($\tan \delta$) of ceramics corresponding to different compositions. (a) Curves at several frequencies during heating for composition 1 (arrows indicate increasing frequency, T_c - Curie temperature). (b,c) Curves at 1 MHz during a heating/cooling cycle for compositions: 1 - Bi_{0.62}Pb_{0.38}Fe_{0.52}Mn_{0.1}Ti_{0.38}O₃, 2 - Bi_{0.62}Pb_{0.38}Fe_{0.47}Mn_{0.15}Ti_{0.38}O₃ with quenching, 3 - Bi_{0.58}Pb_{0.42}Fe_{0.47}Mn_{0.11}Ti_{0.42}O₃, and 4 - Bi_{0.75}Pb_{0.25}Fe_{0.7}Mn_{0.05}Ti_{0.25}O₃.

Transition temperatures of 520 and 495 °C were obtained from the dielectric curves on heating and cooling, respectively. This is in agreement with the temperature range at which a transition from $Pnma$ to $Pm-3m$ symmetries was described from the temperature evolution of the XRD profiles of slowly cooled powders [Fig. 6(b)]. Note the large difference in T_c found be-

tween the heating and cooling runs ($\Delta T_C \sim 35$ to 25°C), an effect characteristic of materials at MPBs, and that has also been related to different sequences of polymorphic phase transitions during heating and cooling runs.^{14,22} Indeed, related effects have been clearly identified in quenched ceramics, as it is illustrated for composition 3 in Fig. 8(c). Recall that a tetragonal phase had been obtained for quenched powders of this composition [Fig. 3(c)]. It is remarkable the highly irreversible character of the transition during a first thermal cycle, with a maximum relative permittivity in excess of 12 000 at T_C of $\sim 490^\circ\text{C}$ on heating, that is, three times greater than that on cooling at $\sim 510^\circ\text{C}$.

Thermal hysteresis of a first-order ferroic transition is always such that T_C on cooling is lower than that on heating. Therefore, results obtained can only be associated to a different sequence of polymorphic transitions, specifically from the $P4mm$ phase on heating, which is the polymorph favored by quenching, and to the Cc phase on cooling (according to the XRD characterization of slowly cooled powders). Also remarkable it is the enhancement of polarizability at T_C in the quenched material, related with the presence of the tetragonal polymorph, as compared to that of the monoclinic phase.

Differences in room temperature phases between samples either slowly cooled or quenched from high temperature have been associated with the key presence of Mn, and the related defect chemistry of these oxides,²⁶ which is usually dominated by A-site and oxygen vacancies along with reduction/oxidation of the magnetic species at the B-site.^{36,40} One can expect the equilibrium oxygen content at high temperature to be frozen at room temperature by quenching, favoring the tetragonal distortion. A slow cooling, however, provides more time for equilibration of the oxygen, involving variations in the oxidation state of cations at B-sites (Mn and Fe), thus resulting in changes in the tolerance factor and pushing the system across the MPB. Specifically, the antipolar orthorhombic $Pnma$, which is one of the most common space groups in perovskite oxides,⁴¹ has been related to the presence of Mn^{4+} and the resulting decreased distortions that it produces, favoring the centrosymmetric structure.^{26,36} Note from the schematic phase diagram shown in Fig. 1, that the whole solid solution between BiFeO_3 and PbTiO_3 is polar and ferroelectric, with either $P4mm$, Cc or $R3c$ symmetry. It is the Mn incorporation which results in MPB regions with the antipolar $Pnma$ phase, not only in both binary systems but also in the ternary one.

The role of the oxygen content, and its equilibration kinetics, was confirmed in a series of ceramics of composition 1, with quenching from increasing temperatures. Figure 9(a) shows the temperature dependence of relative permittivity during heating at 1 MHz for the different samples. Only one dielectric anomaly was obtained (at 575°C) in the slowly cooled sample, which must correspond to the ferroelectric to paraelectric transition of the monoclinic polymorph, the one prevailing in this ceramic sample. In the quenched materials, however, a second anomaly appears at $\sim 520^\circ\text{C}$ with increasing the quenching temperature, which largely dominates the dielectric response after quenching at 900°C . This latter anomaly is very similar to that of composition 3 after quenching, as shown in Fig. 8(c). Note again the huge permittivity in excess of 20 000 at T_C , that is, five times greater than that after slow cooling. The anomaly at 520°C thus corresponds to the ferroelectric transition of the tetragonal polymorph to the paraelectric cubic one. Note that the two anomalies can be distinguished in the curve of the ceramic quenched from 800°C , which is likely related to an intermediate situation.

In this case, the oxygen equilibrium at high temperature would only be achieved in the grain shell, so that a reduced shell, then most probably tetragonal, would be formed around a still oxidized core, which is so monoclinic, after the quenching.

Finally, a distinctive grain size effect on the ferroelectric transition was found in ceramics of composition 1, processed at decreasing temperatures. A very similar effect to that found by quenching was obtained when grain size was reduced across the submicron range down to the nanoscale, as shown in Fig. 9(b). Average grain sizes varied between $0.82\ \mu\text{m}$ and $70\ \text{nm}$ in this set of ceramics (Fig. 2). Samples sintered at 1000 and 900°C , with grain sizes in the submicron range, showed a single dielectric anomaly at 575°C , which was related above with the ferroelectric transition of the monoclinic phase, the one prevailing in this material according to XRD. Grain size effect in these two samples was basically a decrease of the maximum

Figure 9. Temperature dependences of the relative permittivity (K') at 1 MHz during heating for ceramics of composition 1 ($\text{Bi}_{0.62}\text{Pb}_{0.38}\text{Fe}_{0.52}\text{Mn}_{0.1}\text{Ti}_{0.38}\text{O}_3$). (a) Samples after thermal treatment at 900°C with slow cooling (SC 900°C) or quenching from increasing temperatures (Q 700°C , 800°C , 900°C). (b) Samples obtained at different hot-pressing temperatures (dashed lines and T_C are Curie temperatures).

permittivity, while no change in T_C was observed. However, the ceramic sintered at 850°C , with a grain size approaching the nanoscale, showed a new dielectric anomaly at a temperature below 500°C , while that associated with the monoclinic to cubic transition is still present at 575°C . Moreover, and in the nanostructured ceramic (the one sintered at 800°C), the latter transition is no longer observed and only one anomaly appears

at ~ 500 °C. These results seem to suggest that tetragonal polymorph is the one stabilized in the nanoscale, just like quenching treatments do. However, this apparent analogy must be handled with care, because oxygen equilibration should be favored by size decrease and thus, one would expect the monoclinic phase to be promoted. Alternative mechanisms need to be considered.

Grain size effects on the functional properties of ferroelectric materials, and in particular those with MPB, have been extensively studied, and an understanding of the mechanisms that rule these size effects has been built.⁴²⁻⁴⁴ The most noticeable effect of decreasing the grain size into the submicron range, and further down to the nanoscale, on the dielectric properties of ferroelectric ceramics is the progressive broadening and depletion of the maximum in permittivity at the phase transition. A combination of grain size and grain boundary effects has been proposed, like the disappearance of the complex domain configurations present in coarse-grained ceramics at the nanoscale,⁴⁵ along with the increasing contribution to the dielectric response of the large volume fractions of grain boundaries with strongly reduced permittivity.⁴³

Grain boundary effects could be behind the decrease in permittivity at the monoclinic to cubic transition with reducing grain sizes in the submicron range (samples sintered at 1000 and 900 °C). However, the new anomaly emerging at 500 °C for nanostructured samples must have a different origin. As mentioned above, this seems to correspond to a different polymorph that might be either tetragonal or orthorhombic, which are the other two phases involved in this region where two MPBs meet (see Fig. 1). We have argued that a tetragonal polymorph would not be expected from oxygen equilibration arguments. Besides, general trend in tetragonal perovskites is the final disappearance of the tetragonal distortion with the size decrease, once ferroelastic domains are not formed at the nanoscale.^{45,46} Therefore, the antipolar *Pnma* phase seems to be the stabilized phase. This might be the consequence of a more complete sample oxidation as a result of the decreased diffusion distances, so that a sequence of *P4mm*, *Cc* and *Pnma* would take place in this composition when increasing levels of Mn^{4+} are introduced. Actually, a monoclinic to orthorhombic transition can be promoted in $\text{BiMnO}_{3+\delta}$ by increasing δ .⁴⁷ However, a true intrinsic size effect cannot be ruled out.

4. CONCLUSIONS

We have prepared a number of compositions belonging to the BiFeO_3 - BiMnO_3 - PbTiO_3 ternary system at or next to a line of MPBs, in the region where the evolution from *P4mm/Cc*-type to *P4mm/Pnma*-type MPBs takes place, to deepen into the correlations among the crystal structure, defect chemistry and physical properties of the three polymorphs involved in phase coexistence within this region. High-resolution XRD diffraction allowed establishing the actual symmetry of the phases coexisting at the MPB, namely monoclinic *Cc*, tetragonal *P4mm* and orthorhombic *Pnma*, so that up to three polymorphs might be present in the same material. Strong thermal history effects in the phase coexistence were found, so percentages of coexisting phases significantly change whether the material is quenched or slowly cooled from high temperature. The origin of this behavior is related with the defect chemistry and the role of mixed-valence states at *B*-sites.

The overall magnetic behavior differs significantly for samples with the different thermal treatments, allowing establishing the type of magnetic interactions for each polymorphic phase.

A weak ferromagnetic moment was found at room temperature within the *Cc* phase, which is also ferroelectric and allows lattice transverse softening. Linear magnetoelectric responses can be expected in this phase. On the contrary, near-paramagnetic behavior down to very low temperatures was found for the *P4mm* and *Pnma* polymorphs, whereas the former was ferroelectric (and highly polarizable) and the latter antiferroelectric (and so antipolar by nature). Large phase-change magnetoelectric responses can be then anticipated in this region as a result of field-induced phase transitions. Besides, high-quality ceramic materials with tailored microstructures, and average grain sizes across the submicron range and approaching the nanoscale, were processed by hot-pressing of nanocrystalline powders obtained by mechanosynthesis, which enabled additionally addressing size effects. Distinctive effects of grain size on the transition sequence, and thus on the phase stabilized at room temperature, were found when size was reduced down to the nanoscale. This provides an additional degree of freedom to tailor phase coexistence, and so to obtain controlled percentages of polymorphic phases with differentiated magnetism and ferroelectricity in the same material for enhanced phase-change magnetoelectric responses.

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Notes

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ABBREVIATIONS

MPB, morphotropic phase boundary; AFM, antiferromagnetic; SRO, spin reorientation.

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